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*LECT-12 V.K.MEHTA CHAPTER-4 (Objective Solution) For UPPCL-JE/AE/PSPCL/UPSSSC-JE/SSC-JE DAILY NEWS*

*ANALYSIS - 13th 14th DECEMBER, 2020 Lect-2*

*V.K.MEHTA ELECTRICAL BOOK SOLUTION* ~~W V Solutions~~

There are many different ways of expressing the concentration of a given solution. Some of the most common include molarity, weight by volume, volume by volume and weight by weight. Weight by volume percent (w/v %) tells you the mass of solute in grams that has been added to a 100 mL solution.

~~How to Calculate w/v (Weight by Volume) | Sciencing~~ **SMALL COMMUNITIES, BIG SOLUTIONS CONFERENCE.**

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A percent w/v solution is calculated with the following formula using the gram as the base measure of weight (w):  $\% \text{ w/v} = \frac{\text{g of solute}}{100 \text{ mL of solution}}$ . Example 1: Physiologic or

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isotonic saline is a 0.9% aqueous solution of NaCl. 0.9% saline = 0.9 g of NaCl diluted to 100 mL of deionized water,

~~Calculating Percent Weight/Volume (% w/v) — LabCE.com ...~~

For example, to find the % w/v of a solution the calculation is: (Mass of Solute (g) / Volume of Solution (ml)) x 100.

Therefore, to figure out the % w/v of a 100ml solution that is made up of 65g nitric acid, we would divide 65g by 100ml and then multiply the answer by 100. This tells us that there is a nitric acid solution of 65% w/v.

~~What Do % V/V, % W/W and % W/V Mean?~~

Tara W. v Yitzchok W. - 2020 NY Slip Op 50219 (U) [\*1] Tara W. v Yitzchok W. 2020 NY Slip Op 50219(U) Decided on February 18, 2020 Supreme Court, Kings County Thomas, J. Published by New York State Law Reporting Bureau pursuant to Judiciary Law § 431.

~~Tara W. v Yitzchok W. :: 2020 :: New York Other Courts ...~~

[\*1] Matter of Jefferson W. v Latonya P. 2018 NY Slip Op 28352 Decided on November 9, 2018 Family Court, Queens County Hunt, J. Published by New York State Law Reporting Bureau pursuant to Judiciary Law § 431.

~~Matter of Jefferson W. v Latonya P. :: 2018 :: New York ...~~

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Because percent solutions can be expressed in three

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different ways, it is imperative that the type of percent solution be explicitly stated. If this information is not provided, the end user is left to "guess" whether w/v %, w/w %, or v/v % was used. Each percent solution is appropriate for a number of different applications.

## ~~Percent (%) Solutions Calculator – PhysiologyWeb~~

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Solutions that require materials from natural sources are often prepared w/v because the molecular formula of the substance is unknown and/or because the substance cannot be described by a single formula. A one percent solution is defined as 1 gram of solute per 100 milliliters final volume.

## ~~Formulas used to describe solutions Rice University~~

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When preparing solutions from dry reagents, the same mass of any reagent is used to make a given percent concentration although the molar concentrations would be different. In general, Weight percent (w/v) = [mass of solute (g)/ volume of solution (ml)] x 100, and, Volume percent (v/v) = [volume of solute (ml)/ volume of solution (ml)] x 100

## ~~Resource Materials: Making Simple Solutions and Dilutions~~

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